

Stellar/Planetary Atmospheres

Part 11: continuous opacities

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Topics

- ▶ continuous opacities
 - ▶ H I
 - ▶ H⁻
 - ▶ metals
 - ▶ electron scattering
 - ▶ molecular opacities

continuous opacities

- ▶ must know all opacities in order to solve RTE and compute spectrum
- ▶ these are functions of T , P_{gas} , P_e , λ , and abundances
- ▶ caution: strictly speaking also of the radiation field!

continuous opacities

- ▶ distinguish between
 - ▶ 'slowly' varying continuous opacities (but discontinuities)
 - ▶ 'rapidly' varying line opacities (dex over 1 Å!)
- ▶ 2 categories for continuum opacities:
 1. b-f transitions: photoionizations
 2. f-f transitions: charged particle (electron) accelerated in E-field of other charged particle (ion) and a photon is absorbed
- ▶ continuous scattering processes:
 1. Thompson (electron) scattering
 2. Rayleigh scattering

stimulated emission

- ▶ typically not included in f-f and b-f transition data!
- ▶ in LTE, it's simply given by

$$(1 - \exp(-h\nu/kT))$$

- ▶ depending on λT , this can be a big correction:
- ▶ $\lambda T = 0.2 \rightarrow \Delta = 0.08\%$ ($\lambda = 2000 \text{ \AA}$ for $T = 10^4 \text{ K}$)
- ▶ $\lambda T = 0.4 \rightarrow \Delta = 2.8\%$ ($\lambda = 4000 \text{ \AA}$ for $T = 10^4 \text{ K}$)
- ▶ $\lambda T = 0.6 \rightarrow \Delta = 10\%$ ($\lambda = 6000 \text{ \AA}$ for $T = 10^4 \text{ K}$)
- ▶ $\lambda T = 0.8 \rightarrow \Delta = 19.8\%$ ($\lambda = 8000 \text{ \AA}$ for $T = 10^4 \text{ K}$)
- ▶ $\lambda T = 1.0 \rightarrow \Delta = 31.1\%$ ($\lambda = 10000 \text{ \AA}$ for $T = 10^4 \text{ K}$)

general stuff

- ▶ opacities written as

$$\kappa(\lambda) = \sum \sigma_i(\lambda) n_i$$

- ▶ n_i from EOS
- ▶ $\sigma_i(\lambda)$: cross section for, e.g., photoionization out of level i
- ▶ $\sigma_i(\lambda)$ from QM or measurements

H I b-f

- ▶ also valid for H I like ions!
- ▶ basic picture:

Table 8.2. Absorption edges.

n	$\lambda, \text{\AA}$	Name
1	912	Lyman
2	3647	Balmer
3	8206	Paschen
4	14588	Brackett
5	22790	Pfund

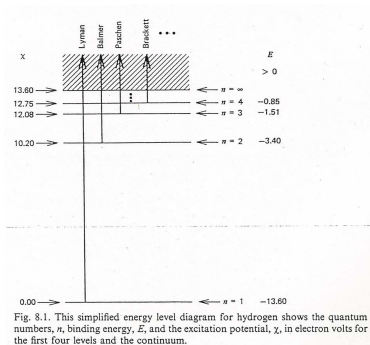


Fig. 8.1. This simplified energy level diagram for hydrogen shows the quantum numbers, n , binding energy, E , and the excitation potential, χ , in electron volts for the first four levels and the continuum.

H I b-f

- ▶ b-f cross section can be computed with QM:

$$\sigma_n(\lambda) = \frac{32}{\sqrt{27}} \frac{\pi^2 e^6}{h^3 c^3} R \frac{\lambda^3}{n^5} g'_n(\lambda) \equiv \alpha_0 \frac{\lambda^3}{n^5} g'_n(\lambda)$$

- ▶ $R = 1.0968 \times 10^5 \text{cm}$: Rydberg constant for H I
- ▶ $\alpha_0 = 1.044 \times 10^5$ (λ in Å)

H I b-f

- ▶ g'_n is the *Gaunt factor*
- ▶ introduced to bring QM result into same form as classical (Kramer's) formula
- ▶ Gaunt factors have been calculated by, e.g., Karzas+Latter, ApJSupp, 6:167 (1961)
- ▶ often used: fit formulae, e.g.,

$$g'_n(\lambda) = 1 - 0.3456(\lambda R)^{-1/3} \left(\frac{\lambda R}{n^2} - 0.5 \right)$$

H I b-f cross sections

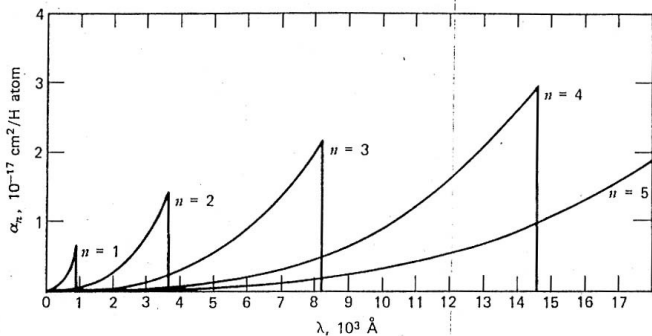


Fig. 8.2. The bound-free absorption coefficient for hydrogen increases with n .

H I b-f

- ▶ \approx small contributions of levels $n > n_T = 3 \dots 6$ by integration:

$$\sum_{n_T}^{n_{\max}} \frac{1}{n^3} \exp(-\chi_n/kT) \approx -\frac{1}{2} \int_{n_T}^{\infty} \exp(-\chi/kT) d(1/n^2)$$

- ▶ with $d\chi = -l_H d(1/n^2)$

$$\begin{aligned} &= \frac{1}{2} \int_{\chi_T}^l \exp(-\chi/kT) \frac{d\chi}{l_H} \\ &= \frac{kT}{2l_H} \left[\exp\left(-\frac{\chi_T}{kT}\right) - \exp\left(-\frac{l_H}{kT}\right) \right] \end{aligned}$$

H I b-f

- ▶ with $\Theta = 5040/T$ this can be written as

$$\kappa_{\text{HIbf}} = \alpha_0 N_H \left[\sum_1^{n_T} \frac{\lambda^3}{n^5} g'_n(\lambda) 10^{-\chi\Theta} + \frac{\log_{10}(e)}{2\Theta I_H} (10^{-\chi_T\Theta} - 10^{-I_H\Theta}) \right]$$

H I f-f

- ▶ f-f absorption is much smaller
- ▶ but important at large λ and/or large T
- ▶ collision electron-proton \rightarrow
- ▶ photon can be absorbed, increasing electron energy
- ▶ probability depends on relative speed
- ▶ classical differential cross section (cm^2 per hydrogen atom):

$$\frac{d\sigma_{\text{ff}}}{dv} = \frac{2}{\sqrt{27}} \frac{h^2 e^2 R}{\pi m^3} \frac{1}{\nu^3 v}$$

H I f-f

- ▶ total cross section by integrating over all electron v 's
- ▶ Maxwell distribution \rightarrow

$$\sigma_{\text{ff}}(\lambda) = \frac{2}{\sqrt{27}} \frac{h^2 e^2 R}{\pi m^3} \frac{1}{\nu^3} \int_0^\infty \sqrt{\frac{2}{\pi}} \left(\frac{m}{kT}\right)^{3/2} \exp\left(-\frac{1/2mv^2}{kT}\right) dv$$

- ▶ so that

$$\sigma_{\text{ff}}(\lambda) = \frac{2}{\sqrt{27}} \frac{h^2 e^2 R}{\pi m^3} \frac{1}{\nu^3} \sqrt{\frac{2m}{\pi kT}}$$

H I f-f

- ▶ QM calculation (Gaunt, 1930) gives basically the same result
- ▶ but adds a f-f Gaunt factor:

$$\sigma_{\text{ff}}(\lambda) \rightarrow \sigma_{\text{ff}}(\lambda)g_{\text{ff}}(\lambda)$$

- ▶ with (λ in Å)

$$g_{\text{ff}}(\lambda) = 1 + 0.3456(\lambda R)^{-1/3} \left(\frac{\lambda kT}{hc} + 0.5 \right)$$

H⁻

- ▶ H⁻ has one *bound* state
- ▶ ionization potential 0.754 eV (16444 Å)
- ▶ not any more bound states!
- ▶ → no spectral lines
- ▶ but important opacity source in the Sun!

H^- b-f

- ▶ polynomial fits to reproduce λ dependence of H^- b-f
- ▶ accuracy better than 0.2%
- ▶ H^- has autoionizing (unstable) levels above its ionization limit
- ▶ \rightarrow resonances in the absorption
- ▶ + forbidden H^- continuum
- ▶ these are very small

H⁻ b-f cross sections

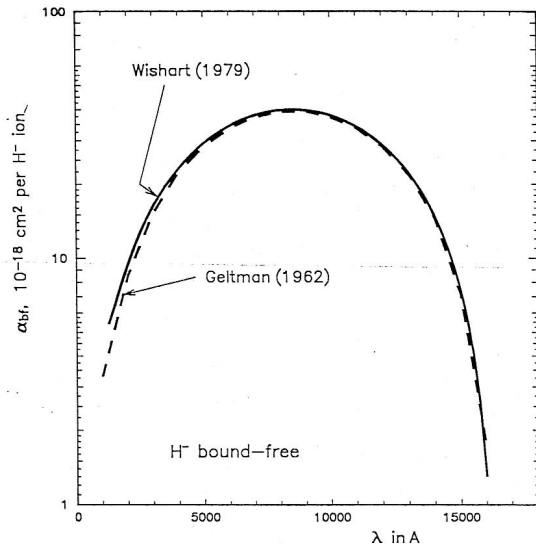


Fig. 8.3. The absorption coefficient of the negative hydrogen ion shows a maximum near 8500 \AA . Two calculations are compared.

H⁻ f-f cross sections

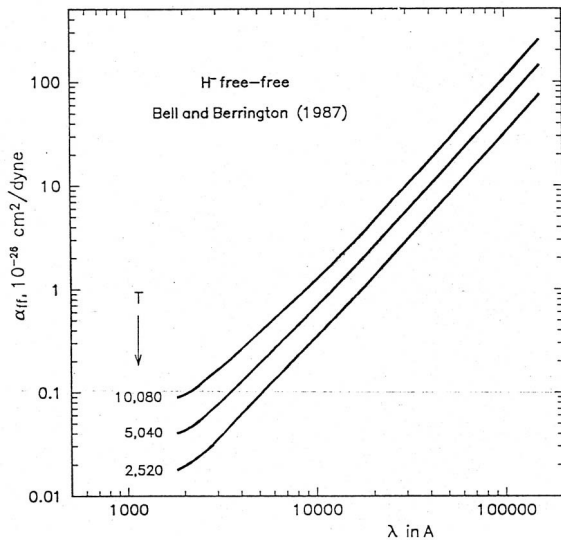


Fig. 8.4. The free-free absorption coefficient of the negative hydrogen ion increases with wavelength.

He⁻ f-f cross sections

Scattering

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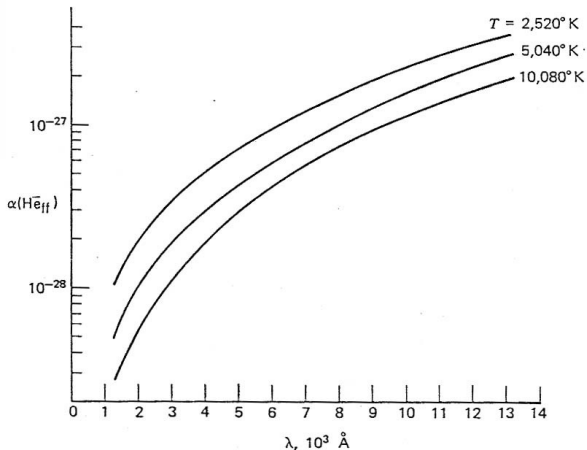
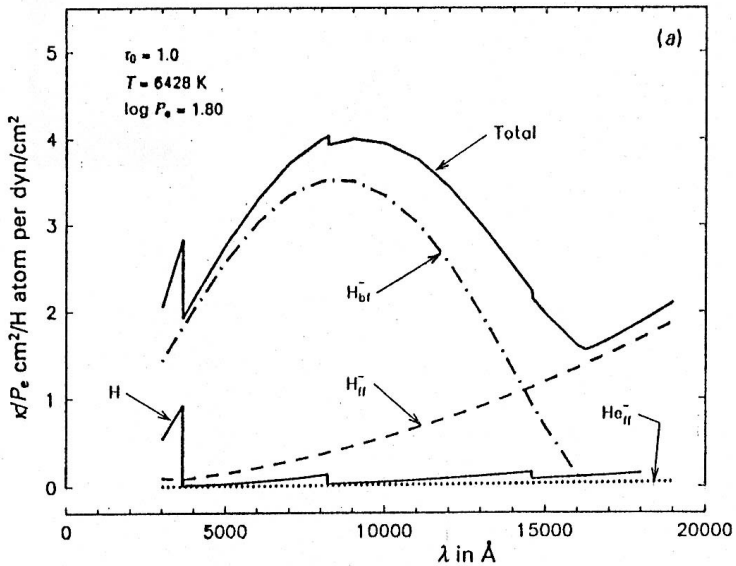
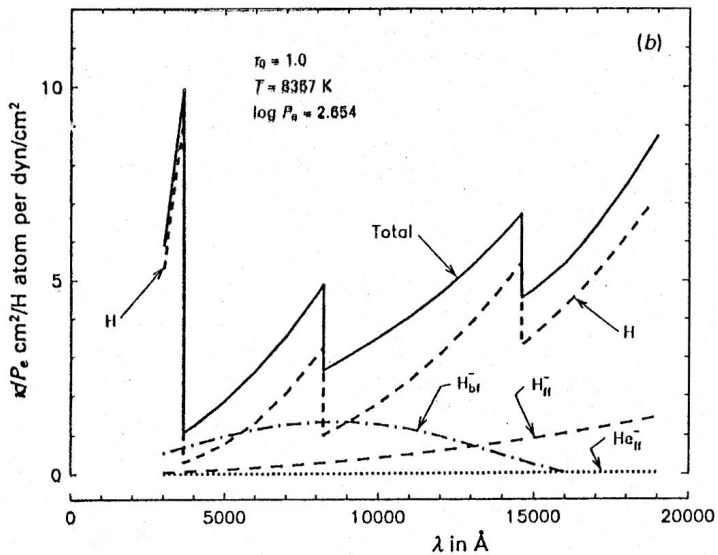


Fig. 8.5. The free-free absorption coefficient of the negative helium ion qualitatively mimics the absorption of the H⁻ ion. The units of $\alpha(\text{He}_{\text{ff}}^-)$ are square centimeters per helium atom per unit electron pressure.

hydrogen opacities



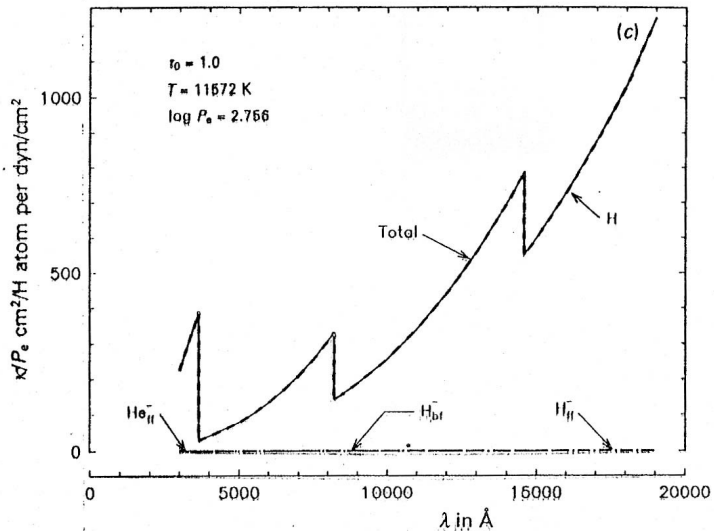
hydrogen opacities



hydrogen opacities

Scattering

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metal b-f opacities

- ▶ metals = everything with $Z > 2$
- ▶ extremely important b-f and f-f opacity sources
- ▶ UV wavelengths!
- ▶ most important elements: C, Si, Al, Mg, Fe
- ▶ few cases: cross sections from lab data
- ▶ most cases: theoretical model calculations
- ▶ 1987: 'opacity project' (Seaton), OPAL (LLNL)
- ▶ → generate state-of-the-art databases of cross-sections
- ▶ assumes LS coupling to use quantum defect methods

electron scattering

- ▶ Thompson scattering: electron scattering in the non-relativistic limit
- ▶ → independent of λ :

$$\sigma_e = \frac{8\pi}{3} \left(\frac{e^2}{mc^2} \right)^2 \approx 0.66 \times 10^{-24} \text{cm}^2$$

- ▶ phase function:

$$\propto 1 + \cos^2 \theta$$

- ▶ → can be approximately treated as isotropic

electron scattering

- ▶ Thompson scattering is important if there are lots of free electrons
- ▶ → high T , small P_{gas}
- ▶ → relatively more important in giants compared to dwarfs

Rayleigh scattering

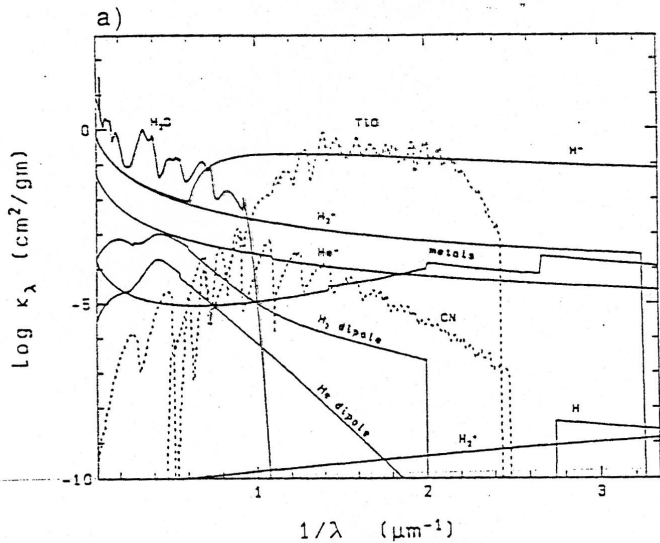
- ▶ important in cool stars
- ▶ important in the UV
- ▶ important for H I, H₂ and He I
- ▶ wavelength dependence of the form

$$\sigma_R(\lambda) \propto \frac{\lambda_0}{\lambda^4} \left[1 + \left(\frac{\lambda_2}{\lambda} \right)^2 + \left(\frac{\lambda_4}{\lambda} \right)^4 \right]$$

molecular opacities

- ▶ cool stars
- ▶ mostly contribute to IR opacity
- ▶ these are spectral lines
- ▶ but *many* of them
- ▶ → overlap to form broad bands
- ▶ most important ones: TiO, H₂O, CO

molecular opacities



molecular opacities

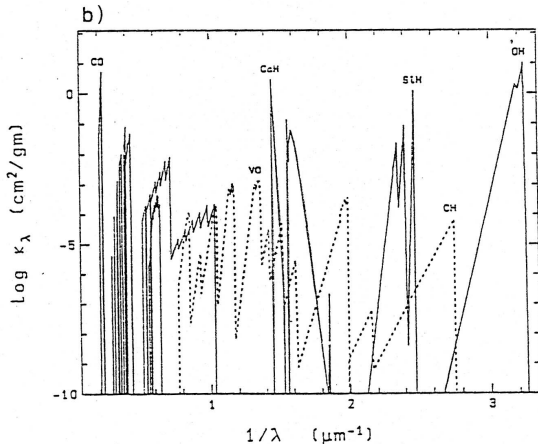
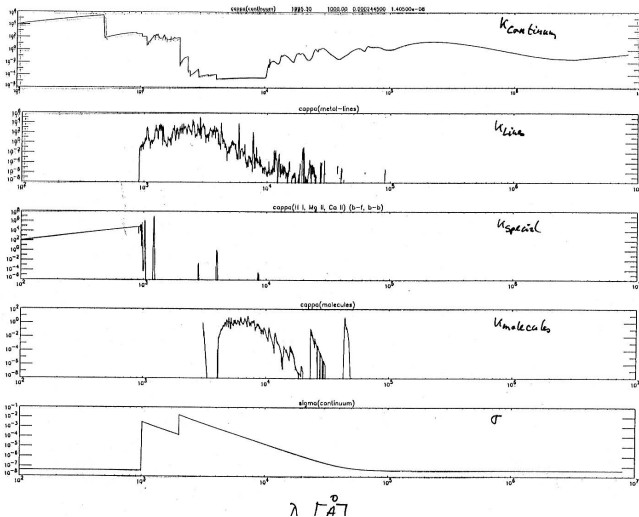


Figure 13: Wavelength distribution of the absorption coefficients in units of cm^2/g obtained at $\tau_{s,10} = 1$ in a model with $T_{\text{eff}} = 3000 \text{ K}$, $\log g = 5.0$ and solar composition for a) the continuum opacity sources (—) and the molecular bands of CN and TiO (- - -) and for b) the molecular bands of the species taken into account in the JOLA (here VO and CH are represented with broken lines for more clarity).

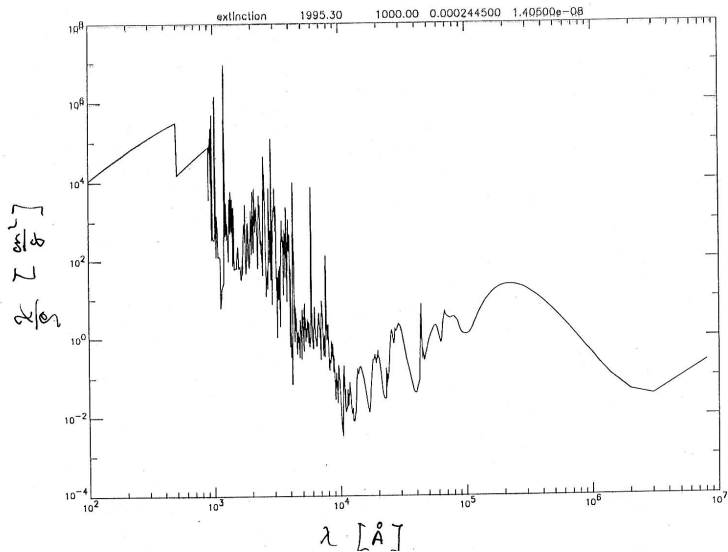
total opacities

$T \approx 20004$, $P_g = 10^3$

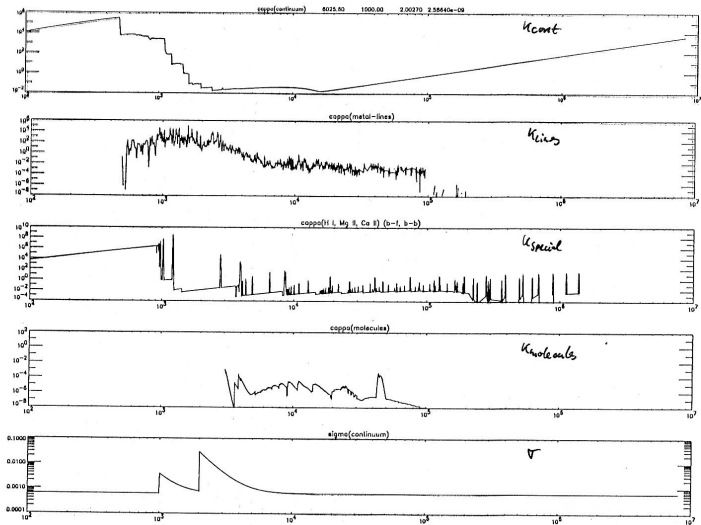
κ [cm²/g]



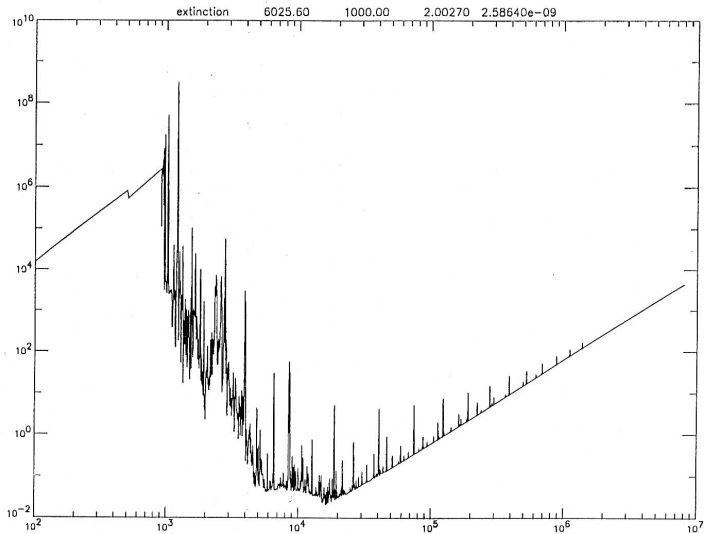
total opacities



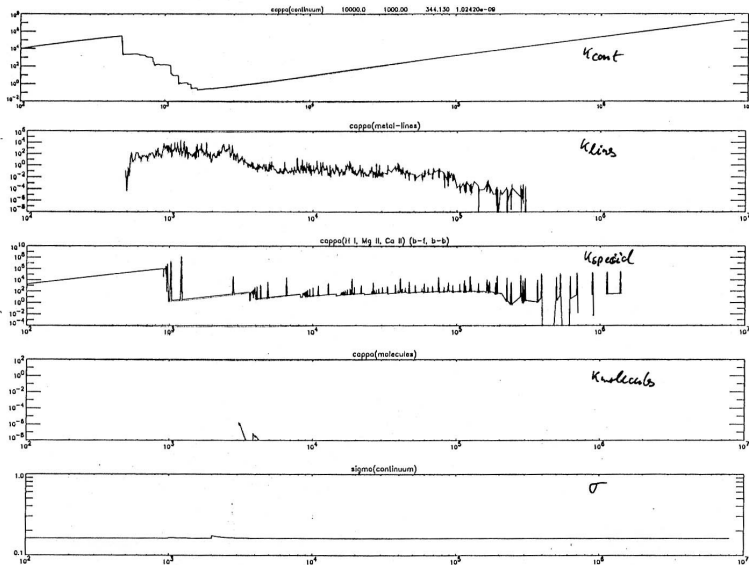
total opacities



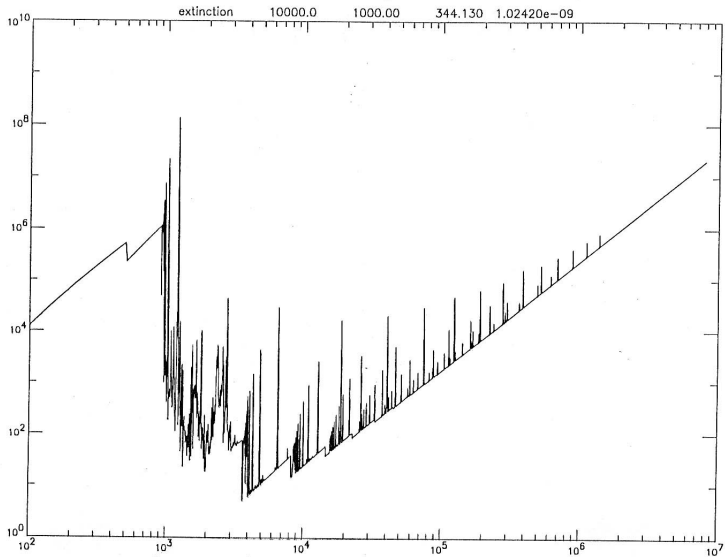
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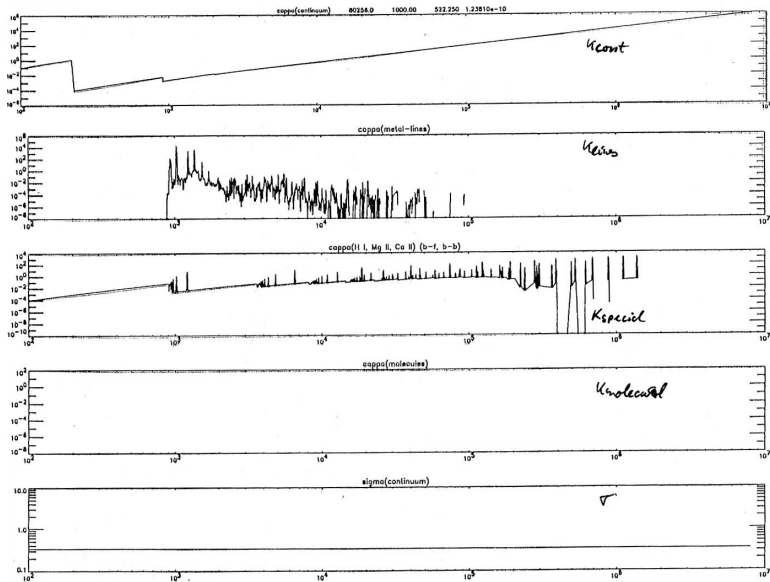
total opacities



total opacities



total opacities



total opacities

